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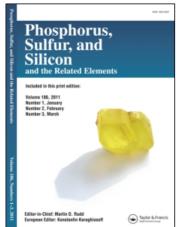
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## Phosphorus, Sulfur, and Silicon and the Related Elements

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## Fluoro-organylphosphanes Using the Reagent Combination $(Et_2N)_3P$ / Fluoro(halo)organic Compound

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We succeeded in a simple direct route with good applicability to transfer different fluoro(halo)organic moieties on chlorophosphanes e.g.: a) the CF<sub>3</sub>-group, b) higher perfluoro-organic substituents, c) the dichlorofluoromethyl-group, d) the pentafluorophenyl-ring and finally e) the trifluorovinyl-moiety.

$$+ Br - CF_{3} \qquad (a)$$

$$+ I - CF_{2}CF_{3} \qquad (b)$$

$$- CF_{2}CF_{3} \qquad (c)$$

$$- (Et_{2}N)_{3}PCIX$$

$$+ CFCI_{3} \qquad (c)$$

$$- (Et_{2}N)_{3}PCIX$$

$$+ Br - C_{6}F_{5} \qquad (d)$$

$$+ Br - CF = CF_{2} \qquad (e)$$

$$+ Br - CF = CF_{2} \qquad (e)$$

In general our reagent combination works very well at low temperatures, where (Et<sub>2</sub>N)<sub>3</sub>P connects phosphorus(III)-halogenides with the fluoro-organic partners under mild dehalogenation. We try to draw your attention to this method for introducing perhaloorganyl groups; the examples give hope to a general scope.